

COOKING WITH THE STARS



What are we made of?



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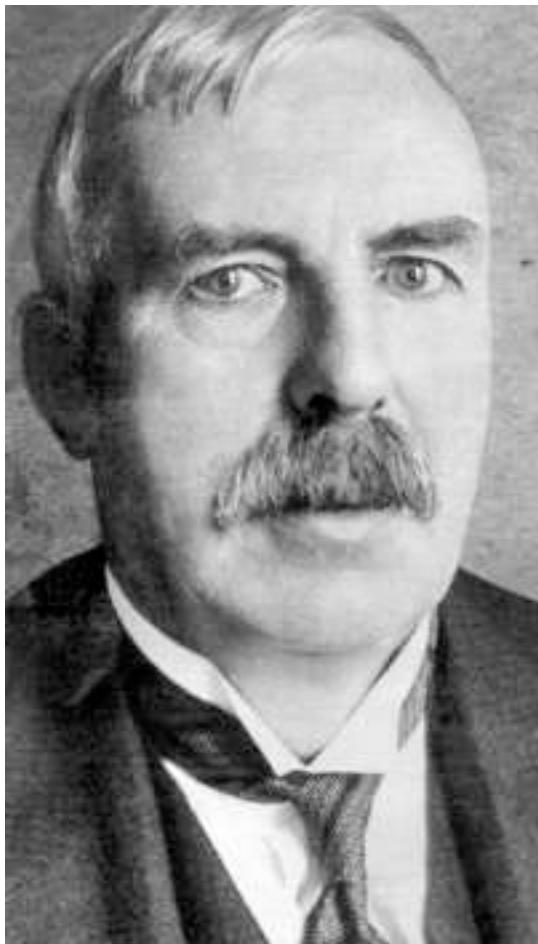
What are we made of?





Element	% (no. of atoms)	How they were made
Hydrogen	61.6	Big Bang
Oxygen	26.3	?
Carbon	9.99	?
Nitrogen	1.48	?
Calcium	0.24	?
Phosphorus	0.20	?
Sulphur	0.06	?
Sodium	0.06	?
Chlorine	0.04	?
Magnesium	0.03	?

Nuclei



Rutherford



Nuclear Safety Officer



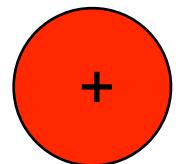
Rutherford's Discovery - 1911



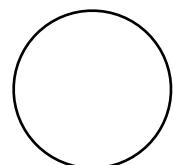
Nucleus

If an atom were as large as the McGill campus, the nucleus would be the size of a dime

Basic Ingredients



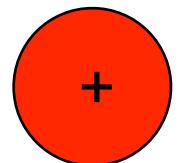
proton



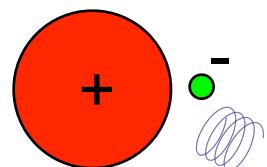
neutron

free protons live forever, but neutrons decay in about 10 minutes

Basic Ingredients



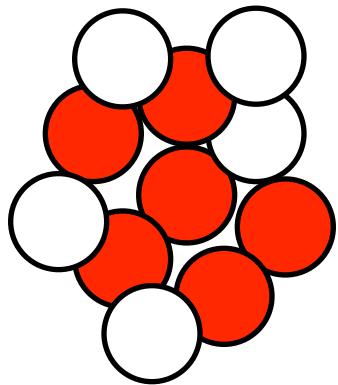
proton



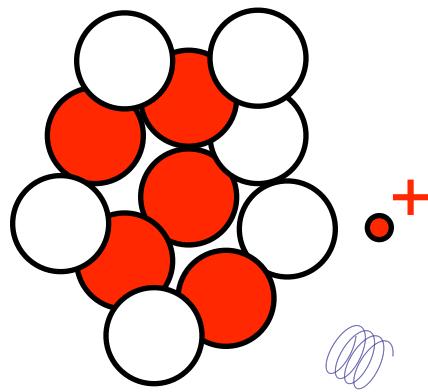
proton

Beta decay of the neutron

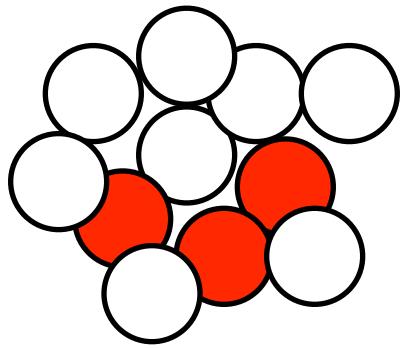
Carbon-II with too many protons



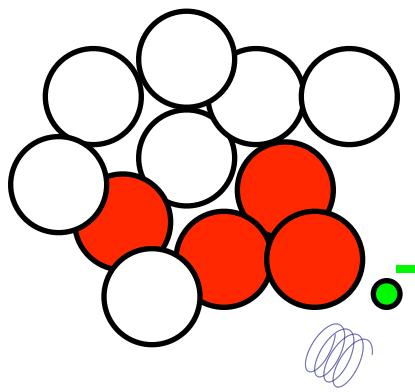
changes to boron-11 by emitting a positron

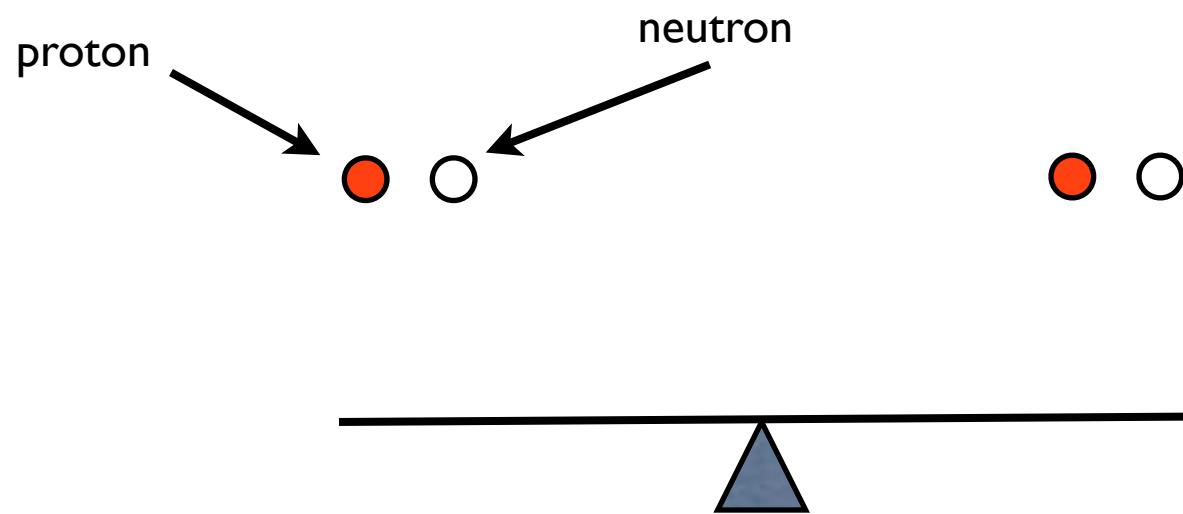


Lithium-11 with too many neutrons

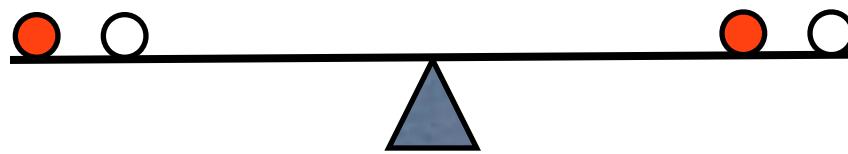


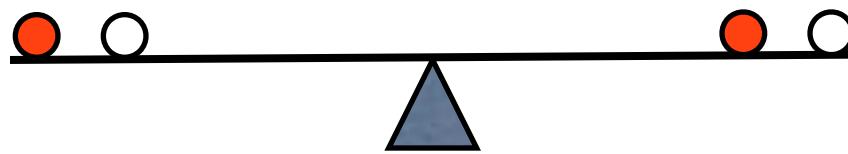
Changes to beryllium - II by emitting an electron



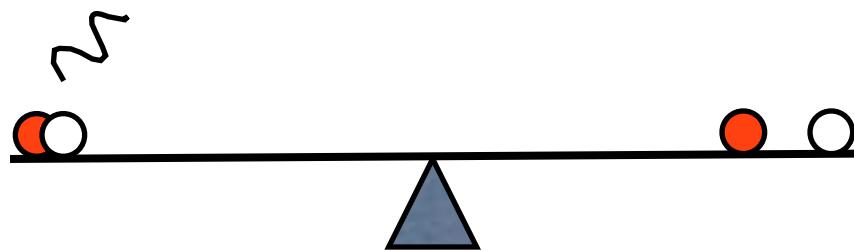


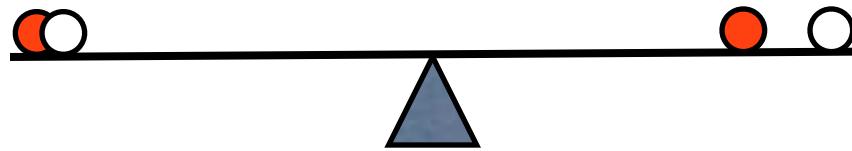
The simplest nuclear reaction: a proton meets a neutron





$$E = mc^2$$

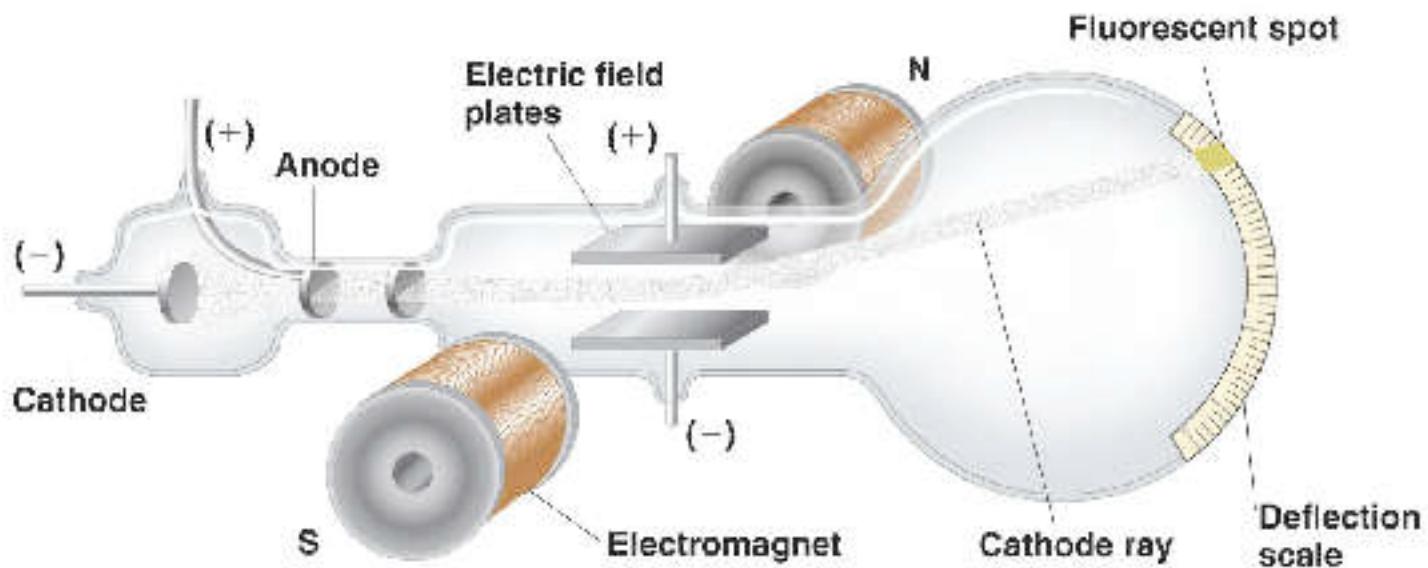




If we could weigh nuclear particles
we could calculate the energy released

J.J.Thompson - 1897

-discovery and mass of the electron-



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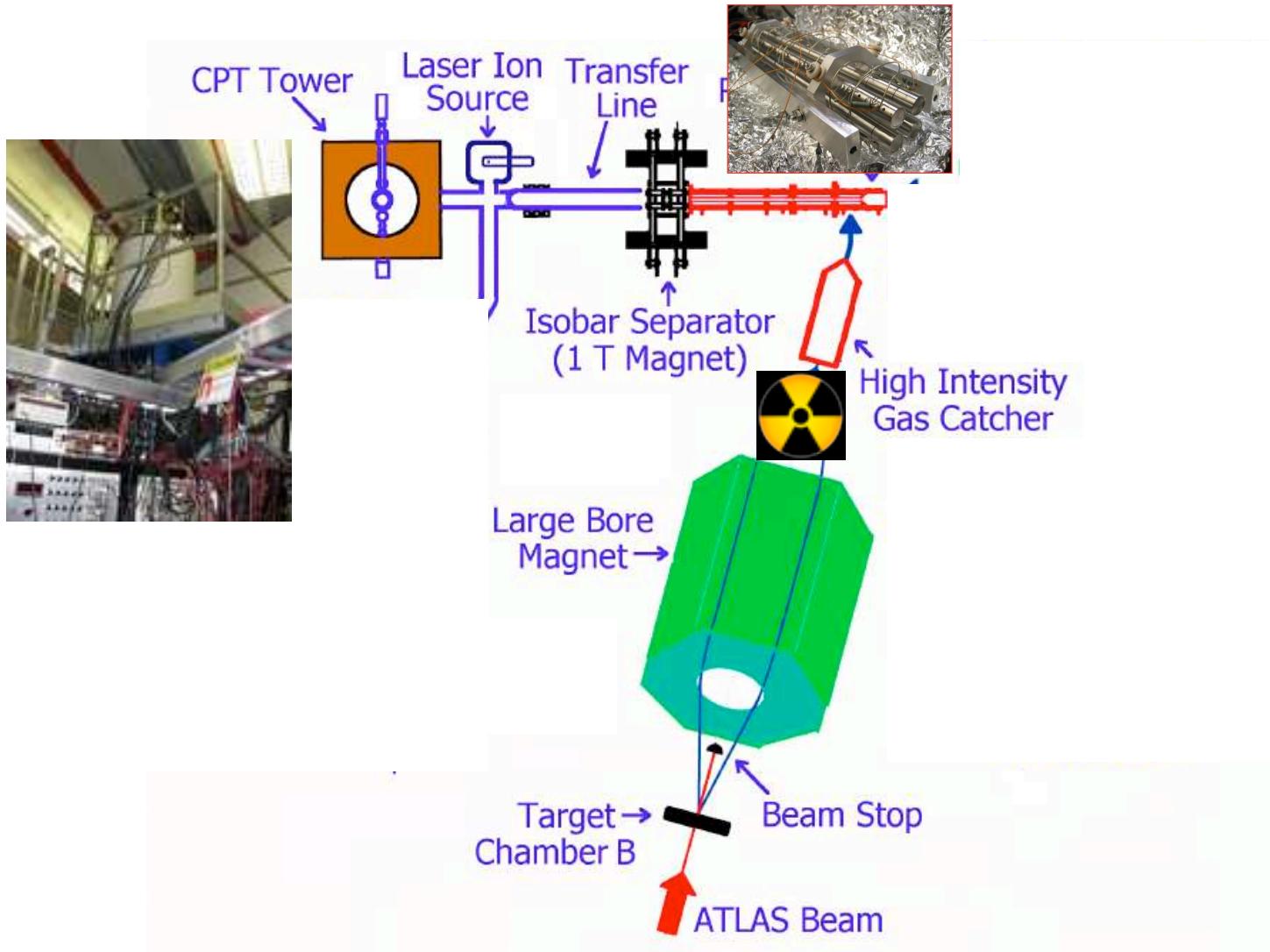
Figure 1: Schematic of J.J. Thompson's experiment.

This was the first mass spectrometer

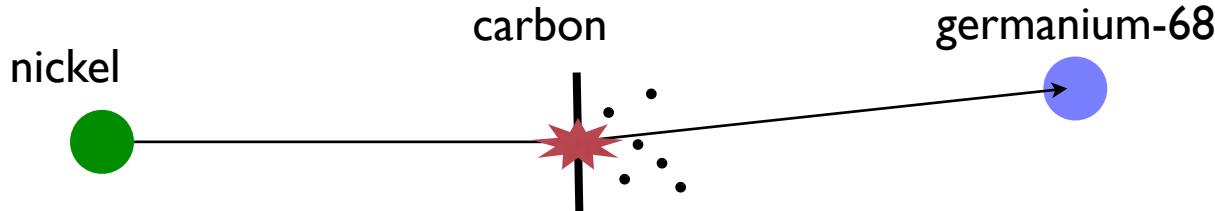
Argonne National Laboratory



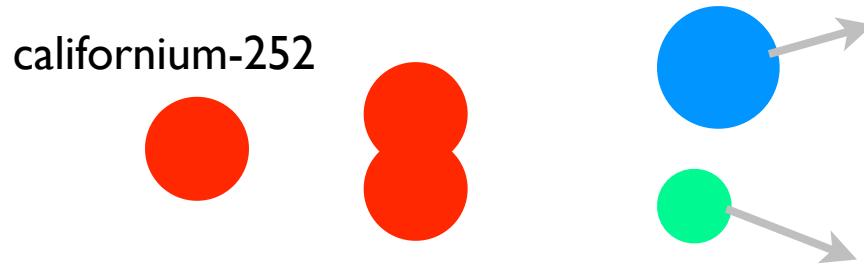
The Canadian Penning Trap



Making proton-rich isotopes by collisions from ATLAS



Making neutron-rich isotopes by californium fission



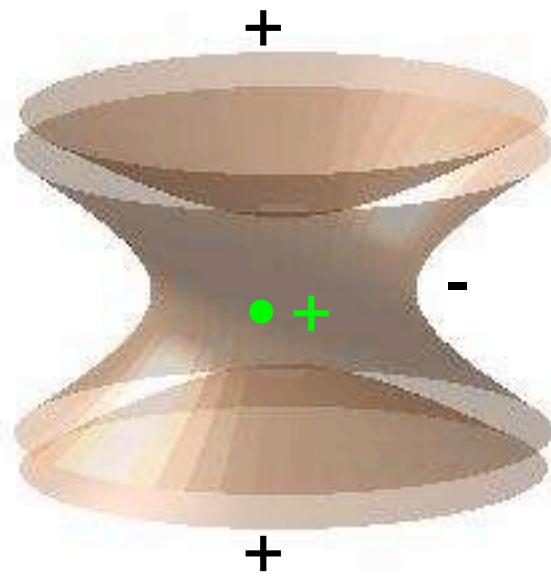
Ion Trap

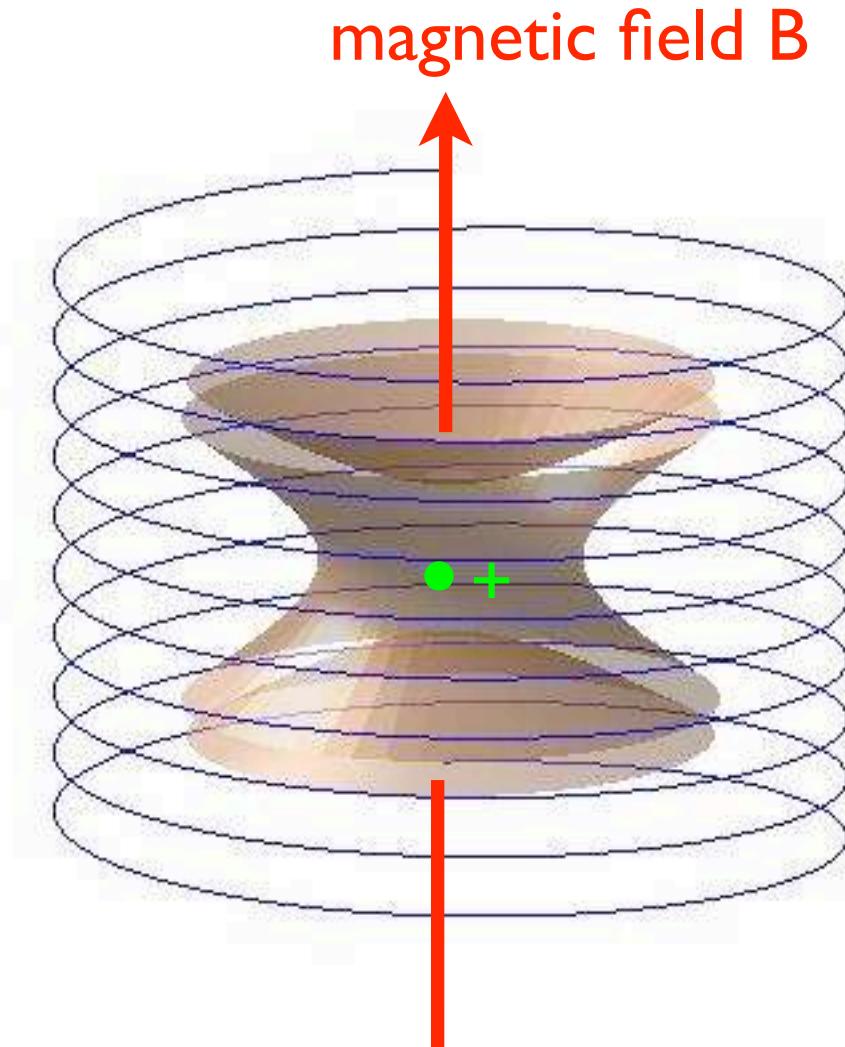


from ion guide



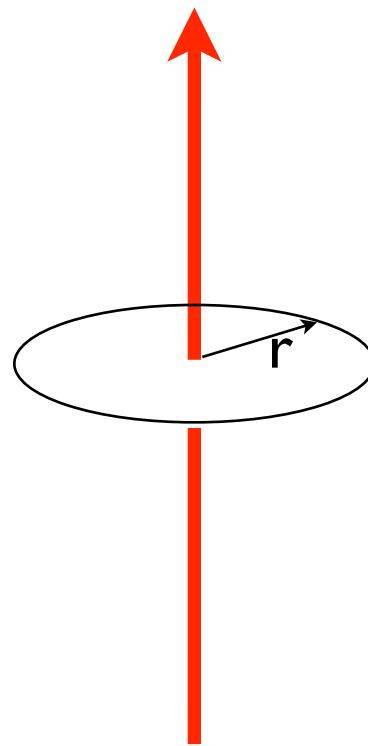
+ve caps keep the ion from escaping axially





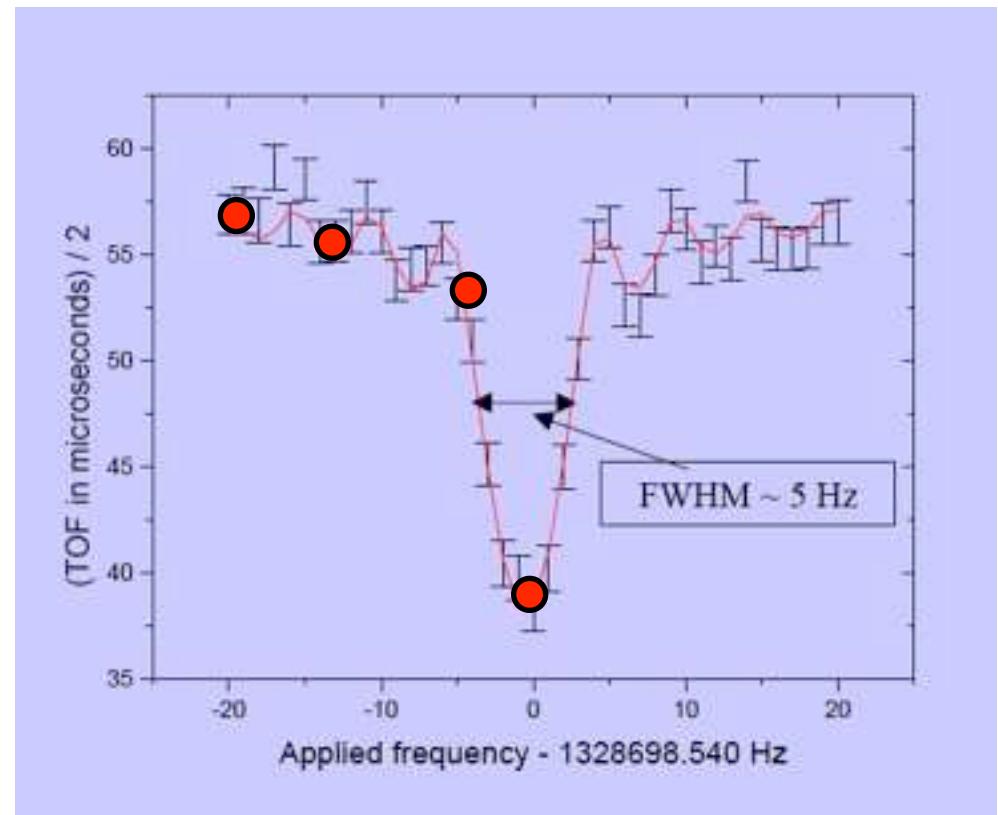
- magnetic field keeps the ion from escaping radially

magnetic field B



rotation speed = constant $\times qB/m$

Measuring the frequency detector



- kick the ions out of the trap
- the ones that have spun in the field arrive first

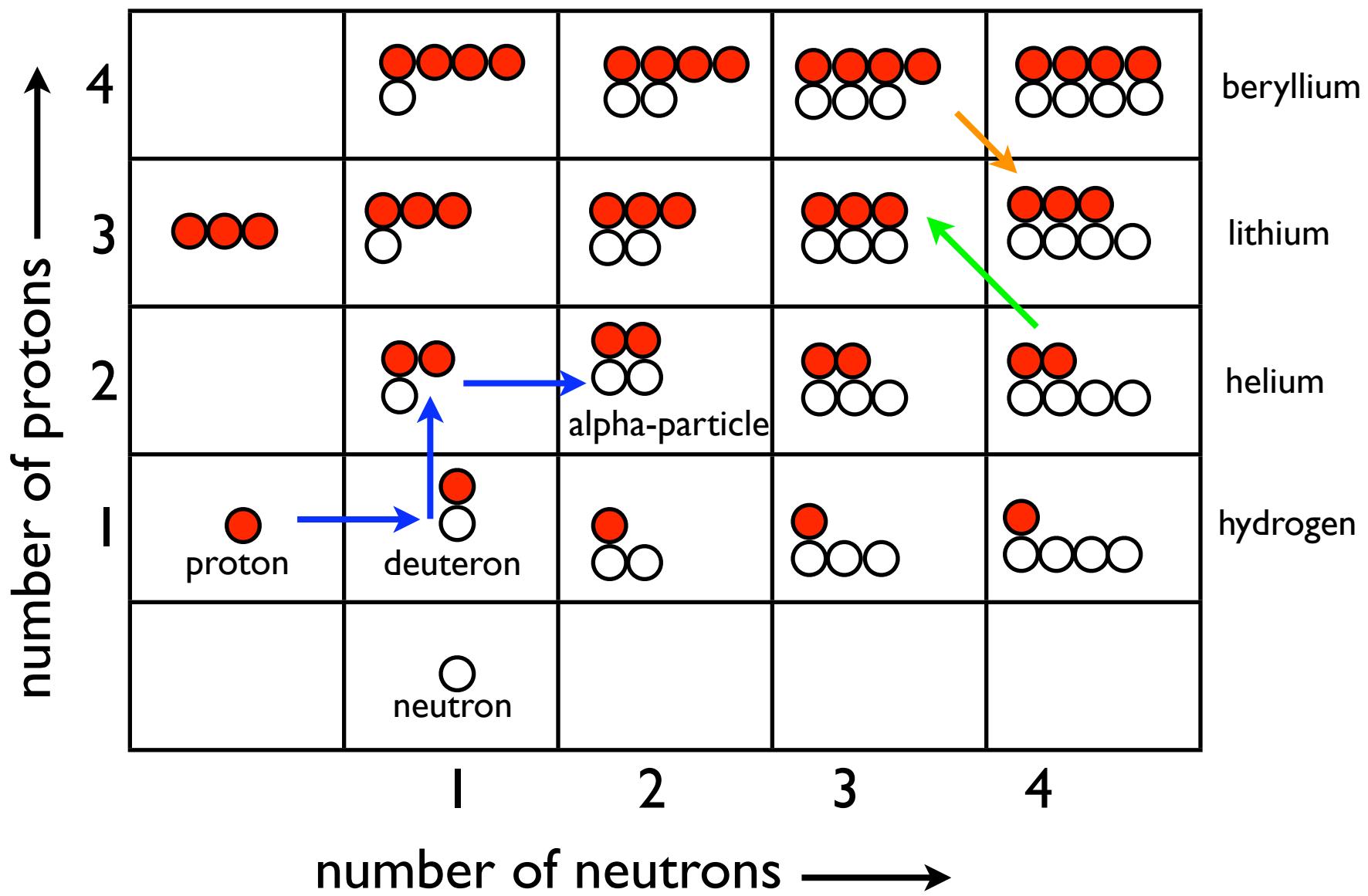
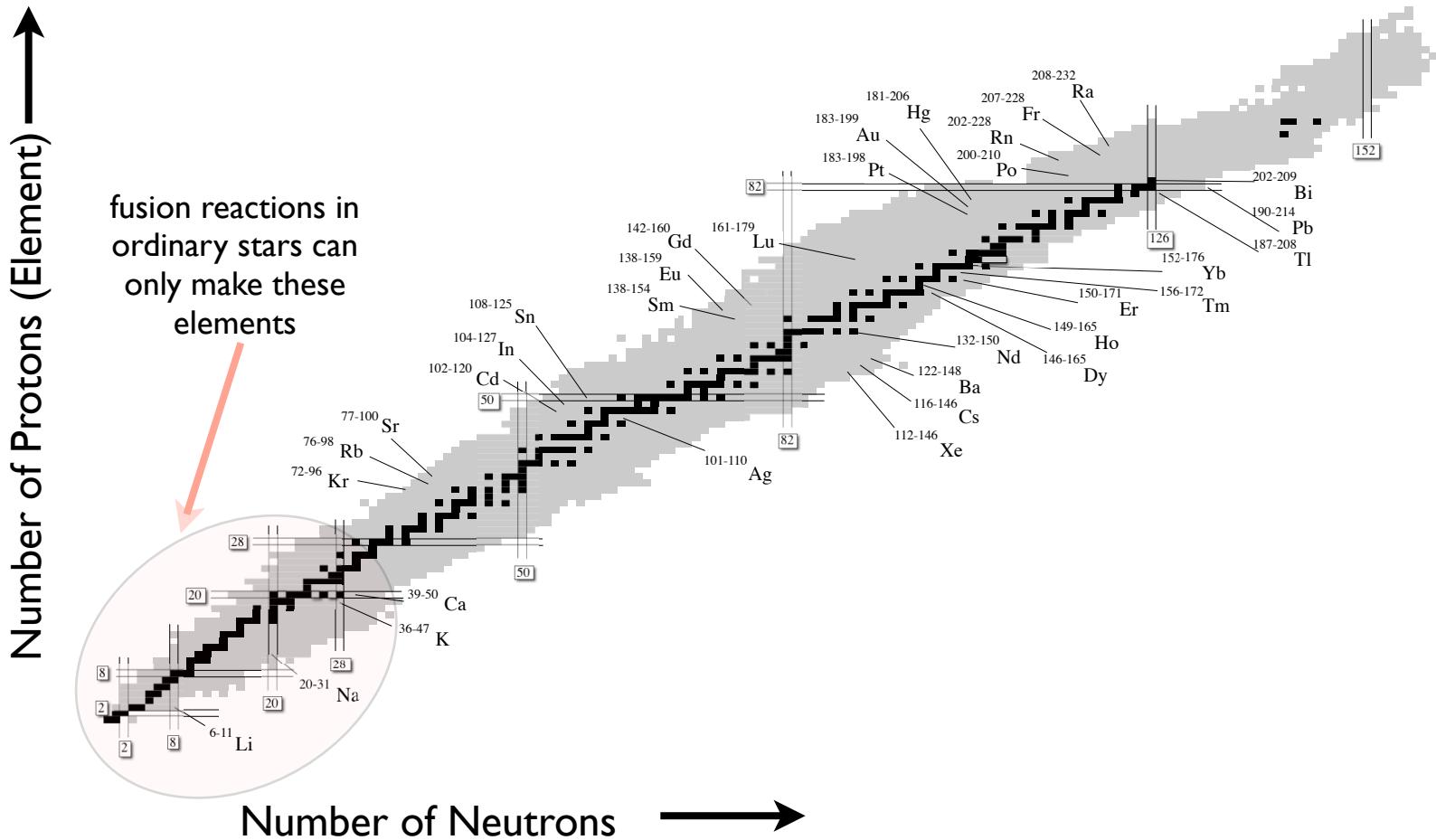
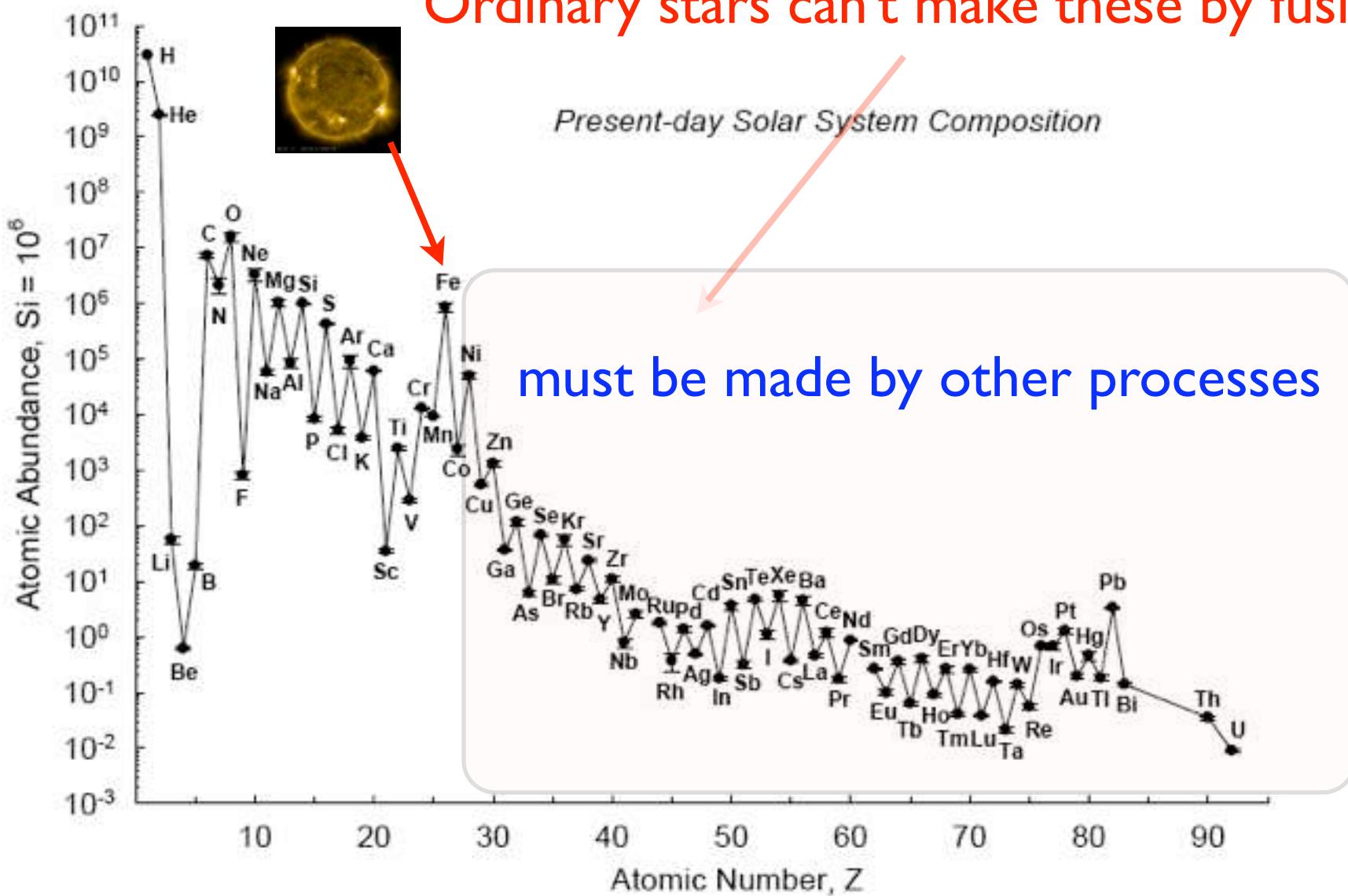
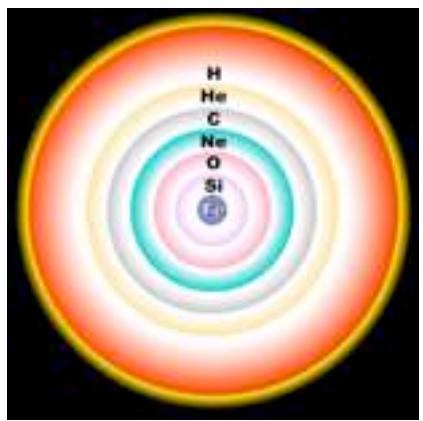


Chart of Nuclides



Ordinary stars can't make these by fusion







Sunday, October 31, 2010

Supernova 1987a - before and after



© Australian Astronomical Observatory

Top left is NE. Width of each image is about 8 arc min

Roll mouse over picture to see arrow (image AAT 50a)

Image and text © 1989-2010, Australian Astronomical Observatory, photograph by David Malin.

Nucleosynthesis in the r-process

JINA

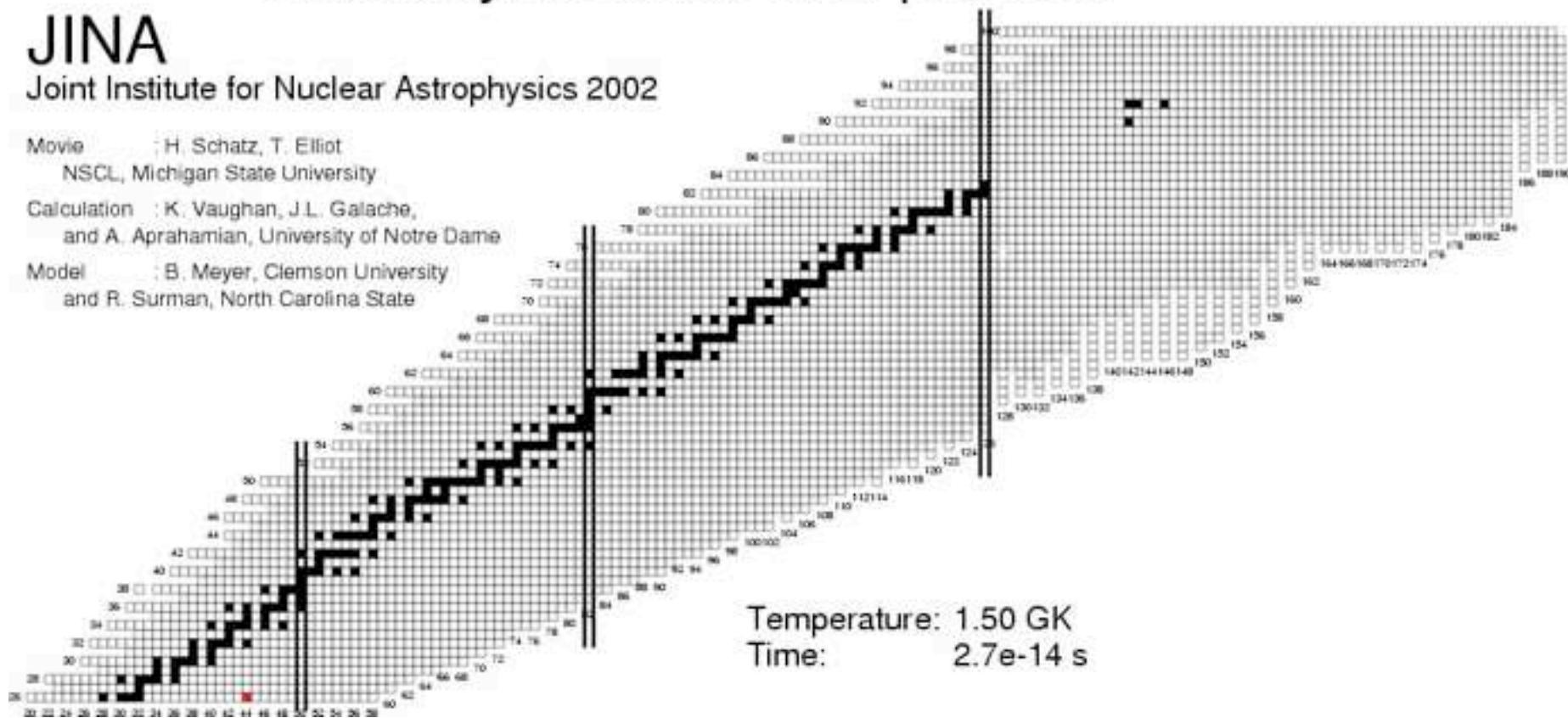
Joint Institute for Nuclear Astrophysics 2002

Movie : H. Schatz, T. Elliot

NSCL, Michigan State University

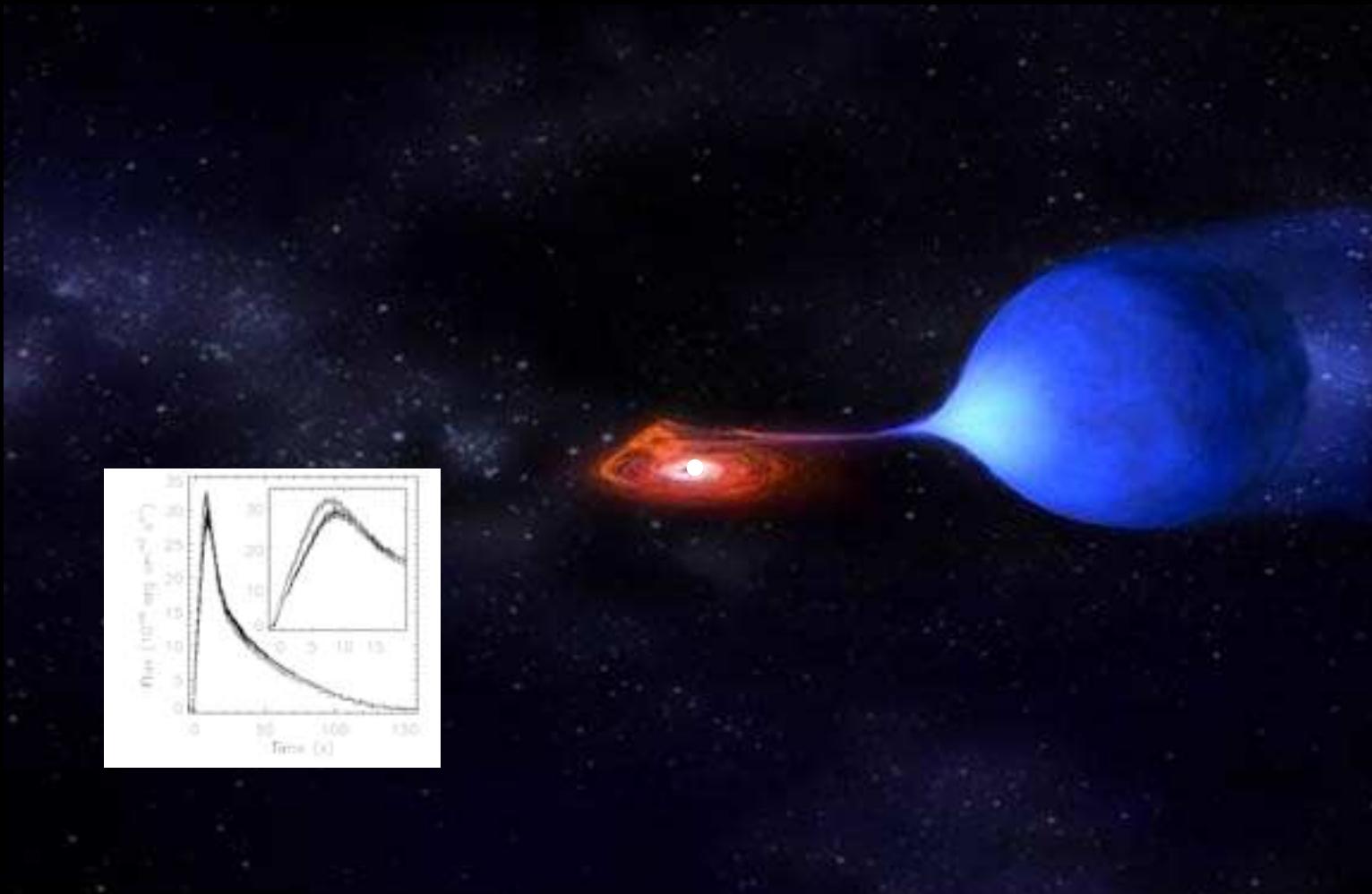
Calculation : K. Vaughan, J.L. Galache,
and A. Aprahamian, University of Notre Dame

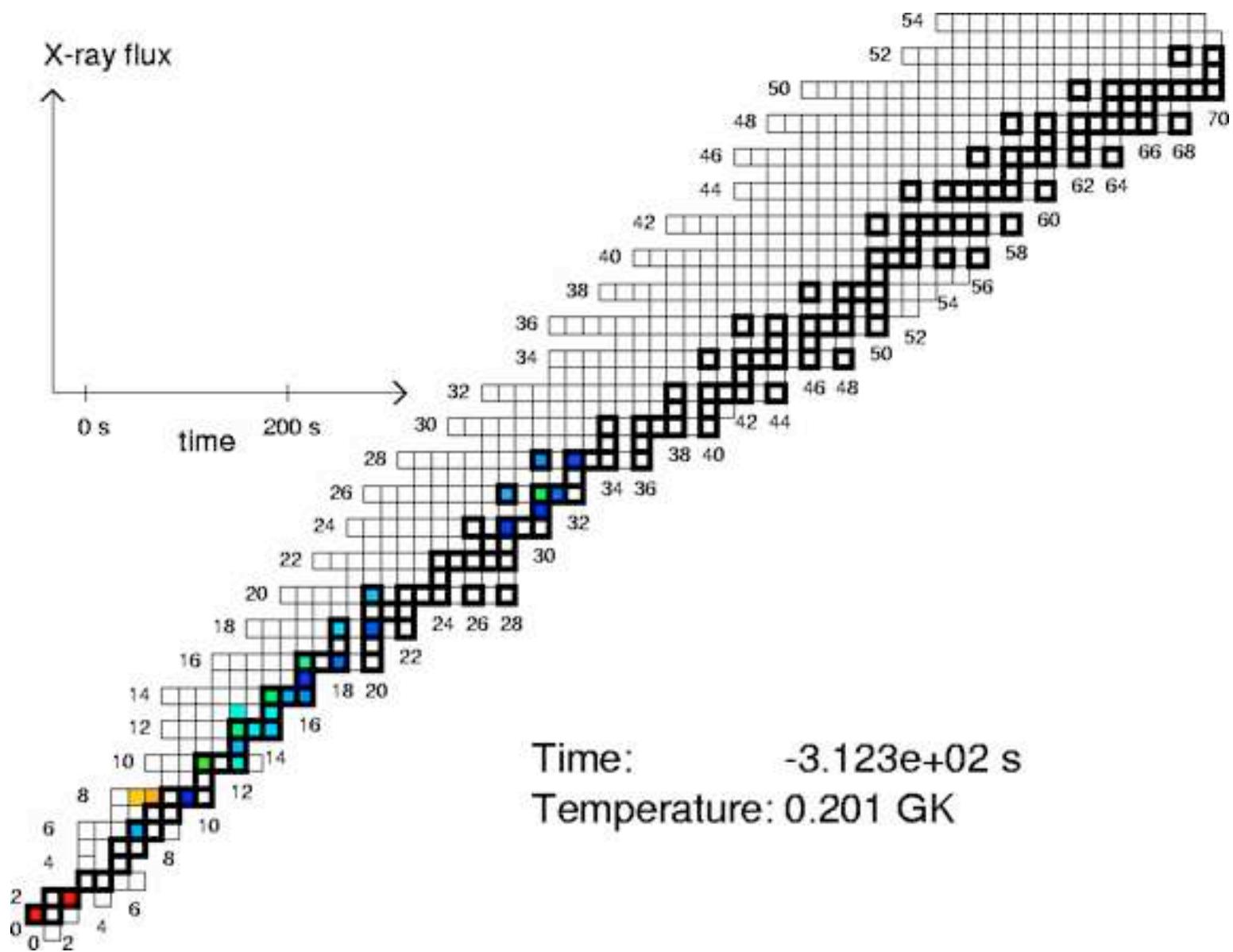
Model : B. Meyer, Clemson University
and R. Surman, North Carolina State



Temperature: 1.50 GK
Time: 2.7e-14 s

X-ray bursts on a neutron star





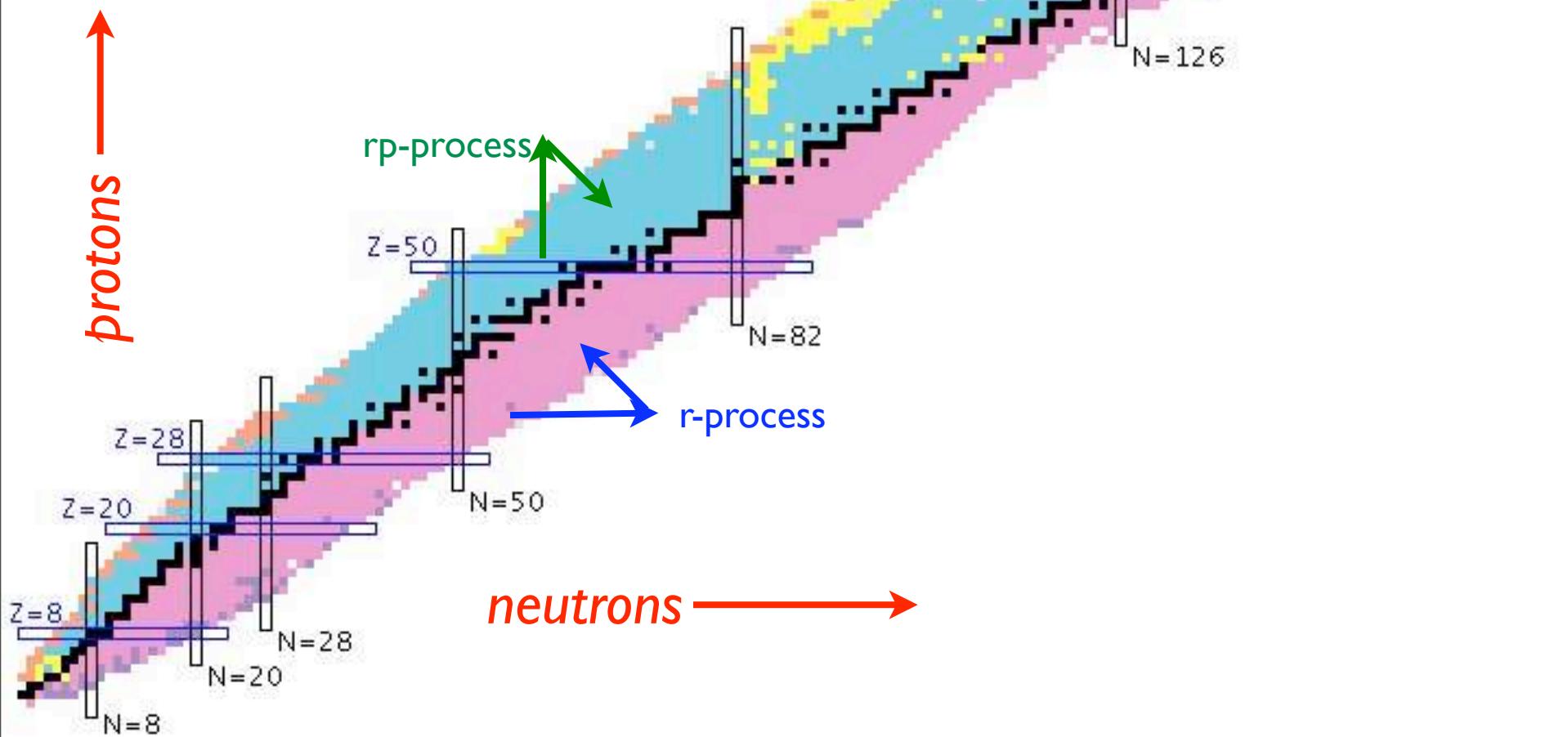
Understanding Stellar Cooking

We need measurements

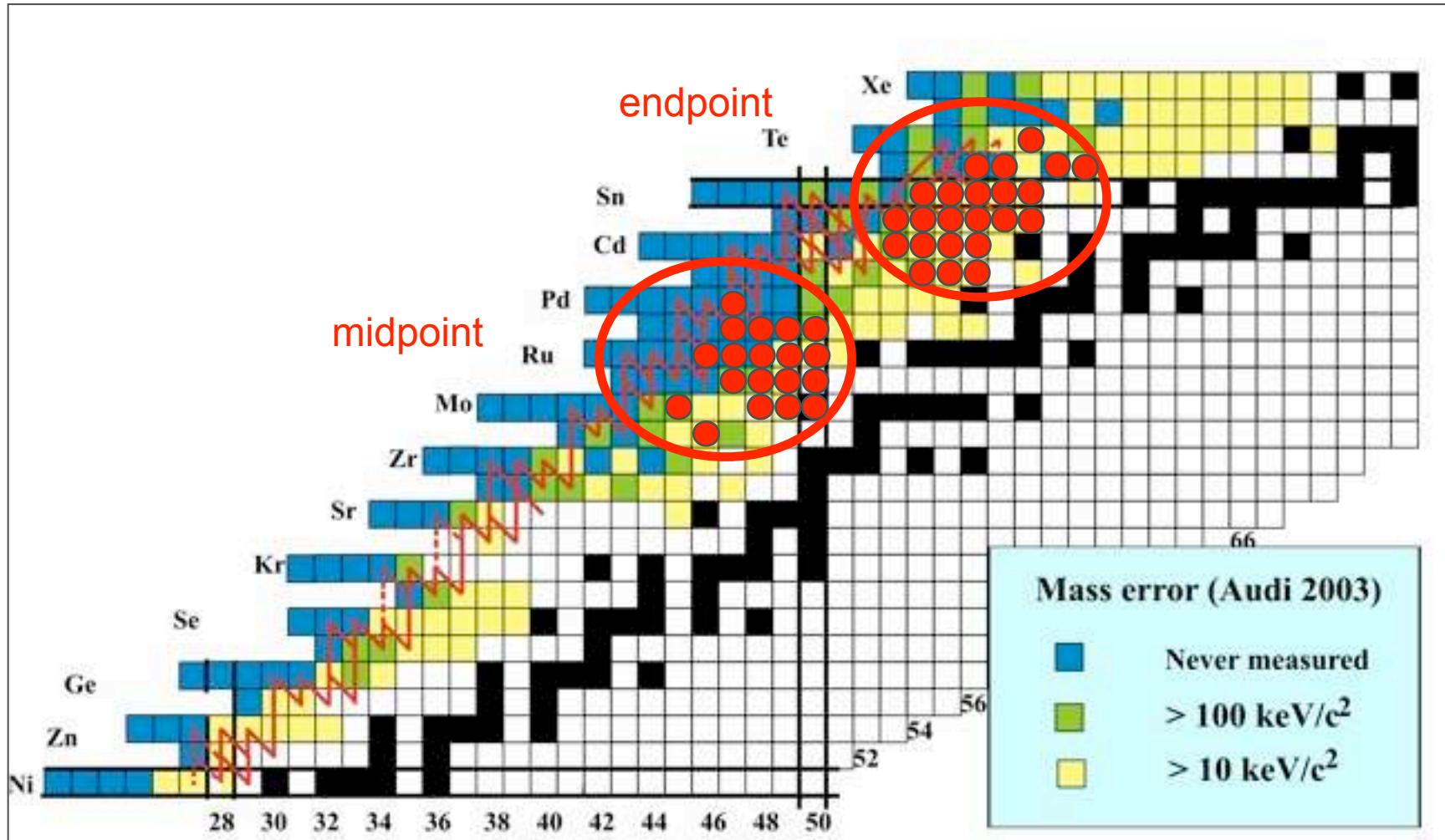
masses for energy differences

decay lifetimes

proton, neutron capture rates



rp-process measurements using the ATLAS beam

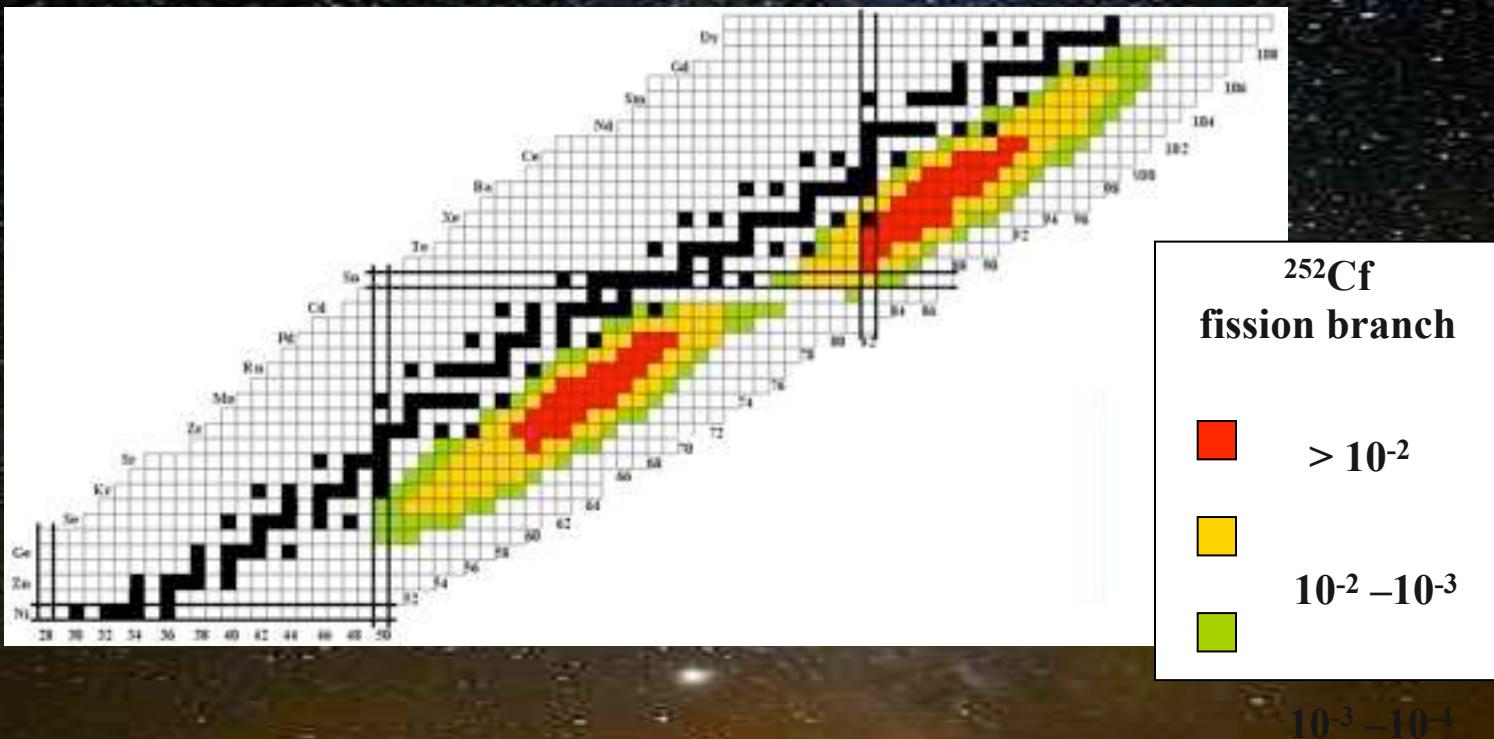


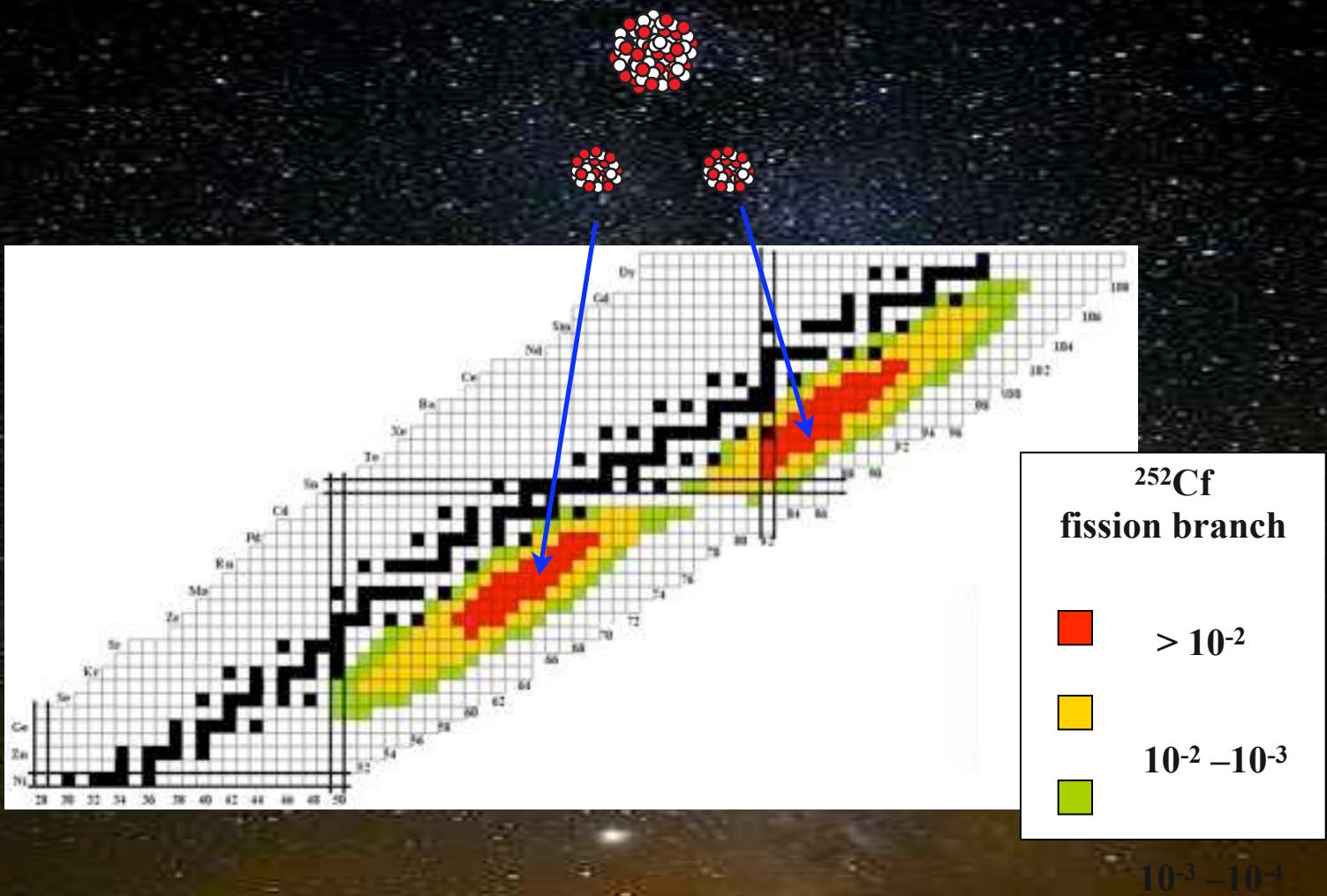
- More than 40 proton-rich nuclides measured over the past few years
- Most masses were determined to better than $10 \text{ keV}/c^2$

G. Audi et al., Nucl. Phys. A 729, 337 (2003).

Californium Fission

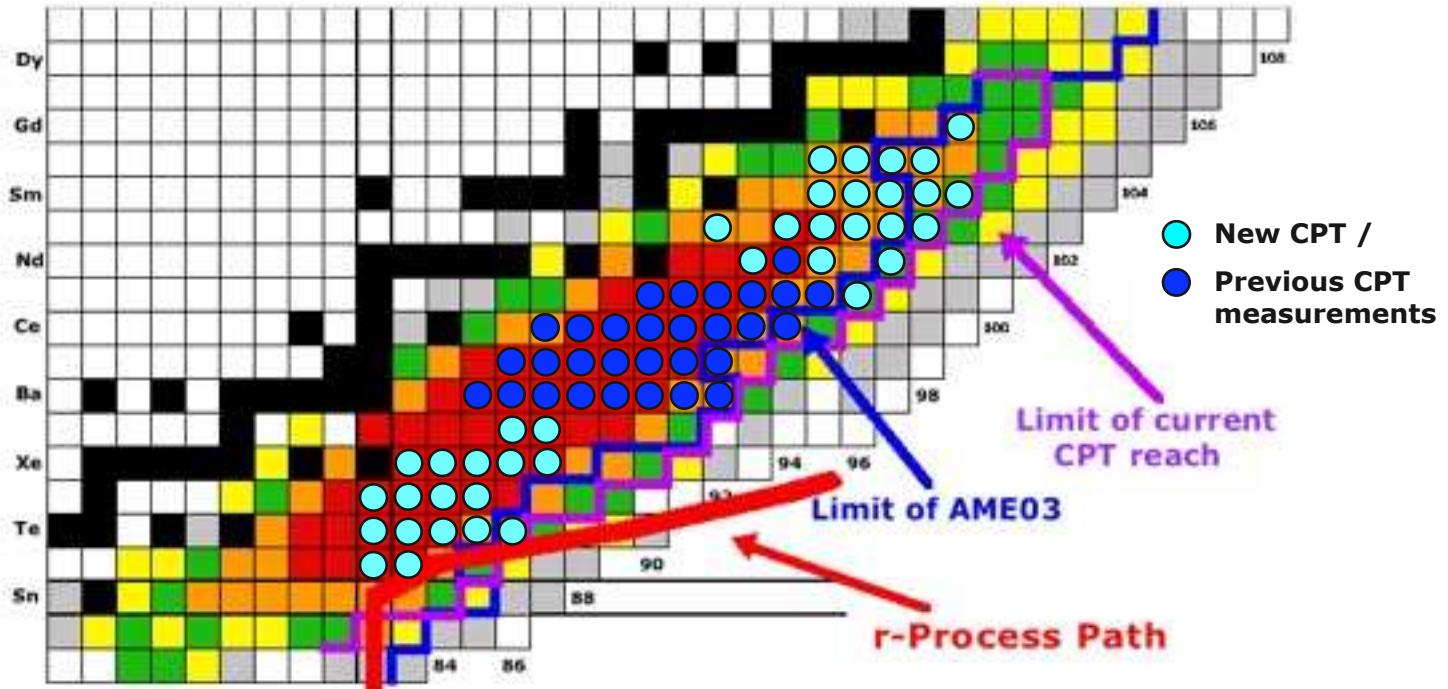
Text





r-process measurements using the californium source

^{252}Cf Heavy Fission Peak



- Ongoing program of measurements since March 2008, target 15 keV uncertainty
- 40 species, 5 have never been previously measured by any means, most others improved by a typical factor of 5
- Adds to 30 measurements taken at CPT in past years with small gas catcher

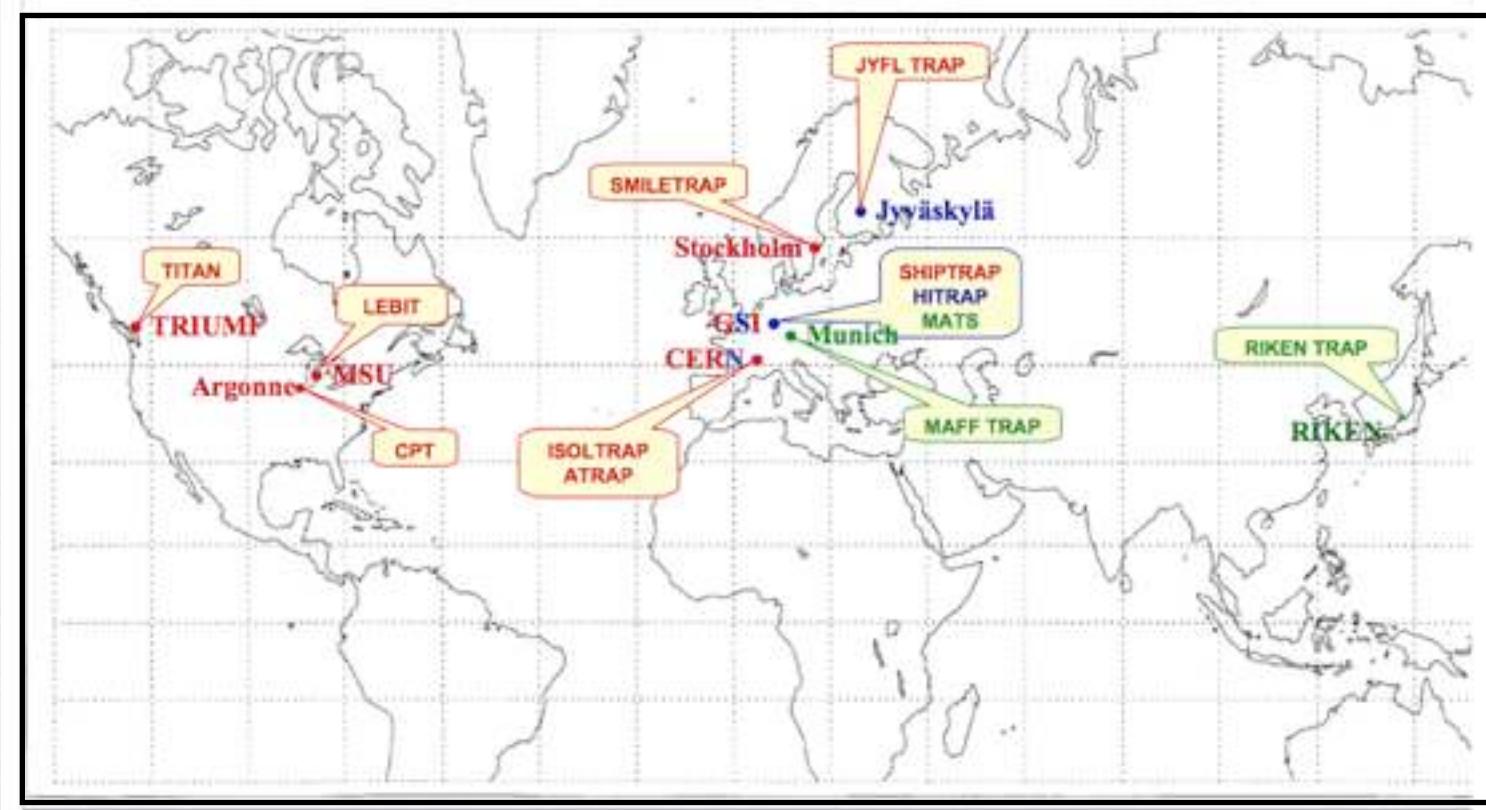


CARIBU

Californium-252 one curie source



Mass Measurement Penning Trap Facilities



Operating

Under Construction

Planned